

Challenges of laser-cooling molecular ions

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Challenges of laser-cooling molecular ions

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Abstract. The direct laser cooling of neutral diatomic molecules in molecular beams suggests that trapped molecular ions can also be laser cooled. The long storage time and spatial localization of trapped molecular ions provides an opportunity for multi-step cooling strategies, but also requires careful consideration of rare molecular transitions. We briefly summarize the requirements that a diatomic molecule must meet for laser cooling, and we identify a few potential molecular ion candidates. We then carry out a detailed computational study of the candidates BH^+ and AlH^+ , including improved *ab initio* calculations of the electronic state potential energy surfaces and transition rates for rare dissociation events. On the basis of an analysis of the population dynamics, we determine which transitions must be addressed for laser cooling, and compare experimental schemes using continuous-wave and pulsed lasers.

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1. Introduction

The long-held notion that laser-cooling molecules is infeasible has recently been overturned by the transverse laser cooling of SrF [1, 2]. The direct Doppler cooling was possible due to the nearly diagonal Franck–Condon factors (FCFs) of the A–X transition and strong optical forces resulting from the short excited-state radiative lifetime. Diagonal FCFs minimize the number of vibrational states populated by spontaneous emission during the cooling time. Similar to previous proposals [3, 4], the number of relevant rotational states involved in the cooling cycle is minimized by judicious choice of the initial angular momentum state [1].

In ion traps, laser-cooled atomic ions can be used to sympathetically cool any other co-trapped atomic or molecular ion species [5–7], and this technique is being used to study gas-phase atomic and molecular ions at very low temperatures in laboratories around the world. For example, sympathetically cooled ions have been used for mass spectrometry [8, 9], precision spectroscopy [10–15] and reaction measurements [16–22]. Because the impact parameter in ion–ion collisions is much larger than the molecular length scale, the internal states of sympathetically cooled molecular ions are undisturbed. A low-entropy internal state can be prepared either by producing the molecular ion through state-selective photoionization [23] or by taking advantage of the long lifetime of the sympathetically cooled molecular ions to optically pump the internal degrees of freedom [24–26]. Now that internal state control of sympathetically cooled molecular ions has been demonstrated, it is interesting to consider the possibility of eliminating the atomic coolant ions (and the accompanying laser equipment) and directly Doppler cooling certain molecular ions. Since molecular ions remain trapped for exceedingly long times independent of internal state, ion traps relax requirements on molecular transition moments and repump rates, thus offering a unique environment for molecular laser cooling. Direct cooling would also be desirable when the coolant ion may create unwanted complications (e.g. quantum information processing using trapped polar molecular ions coupled to external circuits [27]) or when the light driving the atomic cycling transition results in

a change in the molecules' internal state (e.g. resonant absorption or photodissociation). Achieving the necessary cycling of a strong transition would also allow straightforward direct fluorescence imaging of trapped molecular ions at the single-ion level.

On the basis of spectroscopic data available in the literature (see the [appendix](#)), our molecular ion survey found that BH^+ [28–33] and AlH^+ [29], [34–37] are among the most promising candidates. In this paper, we review all the identified challenges in maintaining a closed excitation scheme for the direct laser cooling of BH^+ and AlH^+ stored in ion traps. Unlike in the case of SrF , where slow vibrational decay relative to the interaction time allows a straightforward probabilistic approach to predicting repump requirements [1], designing a laser cooling experiment for trapped ions requires the careful modeling of vibrational decays within the ground state, since they result in the diffusion of parity and rotational quantum numbers. Additionally, for BH^+ and AlH^+ as with all molecular Doppler cooling experiments, rare decay and photofragmentation processes need to be considered, since they can occur on timescales comparable to Doppler cooling and much shorter than the ion trapping lifetime.

This paper is organized as follows. In section 2, we discuss the molecular properties required for Doppler cooling. In section 3, we consider in detail two cooling candidates, BH^+ and AlH^+ , present new quantum chemical calculations for these species (section 3.1.1), discuss the calculation of the Einstein coefficients (section 3.1.2), and present the rate-equation model used to determine the number of cooling photons scattered (section 3.1.3). In section 3.2, we discuss the cooling scheme, including the effect of spin–rotation and spin–orbit splitting of the $^2\Sigma^+$ and $^2\Pi$ states, respectively. The results of our simulation are given in section 3.3, and we present our calculations for rare events, which terminate the cycling transition, in section 3.3.3. In section 4, we discuss the technological requirements for carrying out the proposed experiments, and we propose the use of femtosecond lasers to provide multiple repump wavelengths from a single source. Finally, section 5 summarizes the concepts, simulations and prospects of the Doppler cooling of molecular ions. In the [appendix](#), we present a few additional classes of molecules that need to be studied in more detail to decide whether they are candidates for direct laser cooling experiments.

2. General experimental considerations

Doppler laser cooling requires the scattering of many photons, with each emitted photon on average carrying away an amount of energy proportional to the laser detuning from resonance [38]. To cool a two-level particle with a mass of ~ 10 amu and a visible transition linewidth of ~ 1 MHz from room temperature to millikelvin temperatures requires the scattering of 10^4 – 10^6 photons. Two-level systems are easily obtained in atoms, e.g. ^2P – ^2S transitions in alkali metals (for neutrals) and alkaline earth metals (for ions). In both cases, the ground states have a closed shell with one valence electron, and the state of the atom is well described by the orbital and spin angular momentum of the valence electron [39].

Electronic two-level systems are impossible to obtain in molecules since the vibrational and rotational degrees of freedom introduce multiple decay paths not present in atoms. Spontaneous decay into non-cycling bound or dissociative states generally terminates the cycling transition long before the Doppler cooling limit is reached; however, a carefully chosen molecule can significantly reduce the probability of such decays. The ideal molecule would possess an excited and electronic ground state with equivalent potential surfaces separated by an energy offset in the optical region, resulting in perfectly diagonal FCFs. The most intuitive of the types of

transitions that might lead to diagonal FCFs [40] are transitions that excite a single electron from the highest occupied molecular orbital (HOMO) to the lowest unoccupied molecular orbital (LUMO). This type of transition will have negligible effect on the bond length when both the HOMO and LUMO are either non-bonding or anti-bonding orbitals [41]. Alternatively, one may choose a molecule with an unpaired electron and a corresponding hole in one of the highest bonding orbitals, in which the optical transition moves the hole to another bonding orbital.

In addition to considering the FCFs, one must also take into account the effect of rotational branching due to spontaneous emission, unexpected decay paths due to the breakdown of the Born–Oppenheimer approximation and dipole-forbidden transitions.

2.1. Diagonal Franck–Condon factors

When a decay from the excited state to the ground state occurs, the branching ratio into final vibrational states is determined from the Franck–Condon overlap, which measures the similarity between the excited and ground vibrational wave functions [42]. Molecules that have similar excited-state and ground-state potential energy curves (PECs) reduce the total number of repumping lasers required, so it is useful to determine the repumping requirements as the two PECs are made more dissimilar.

High-accuracy quantum-chemical calculations of the excited PEC are not typically available; however, the Morse potential, with only three parameters, is a reasonable approximation for many bound electronic states. This approximation is particularly true near the bottom of a potential well, which is energetically distant from the region of non-adiabatic couplings that alter the dissociative asymptotes of the potentials. Assuming excited vibrational levels have infinite lifetimes, the number of transitions that are required to approximately close a cycling transition depends strongly on all of the three parameters. Figure 1 shows the effect of changing the equilibrium bond distance and vibrational frequency for fixed anharmonicity. The number of vibrational levels that must be addressed to have less than one part-per-million population loss is very sensitive to differences in the ground- and excited-state bond lengths. In the ideal case of perfectly diagonal FCFs, a single laser is required; a difference of only 5% in the bond lengths requires four to seven lasers.

2.2. Elimination of rotational branching

In the dipole approximation, each emitted photon can change the total angular momentum of the molecule by 0, or $\pm 1\hbar$. The emission results in the population of additional states through Stokes and anti-Stokes processes. Each additionally populated angular momentum state will need to be addressed in order to achieve a high scattering rate and efficient laser cooling.

The number of angular momentum states populated depends on which Hund’s case, (a) or (b), is important [42]. For the purpose of our discussion here, we will initially ignore spin (equivalent to Hund’s case (b)) and later look at complications that occur in Hund’s case (a). We will also ignore nuclear spin and the resulting additional hyperfine states and transitions (the resultant hyperfine splitting is sufficiently small that, in practice, these additional states can easily be addressed by the use of an electro-optical modulator (EOM) [1, 2]).

Λ -doubling breaks rotational symmetry about the bond axis, resulting in electronic states of well-defined symmetry relative to the plane defined by the axis of nuclear rotation and the bond axis. A key feature of the SrF experiment [1, 2] is pumping from the ground state $v' = 0$, $N' = 1$,

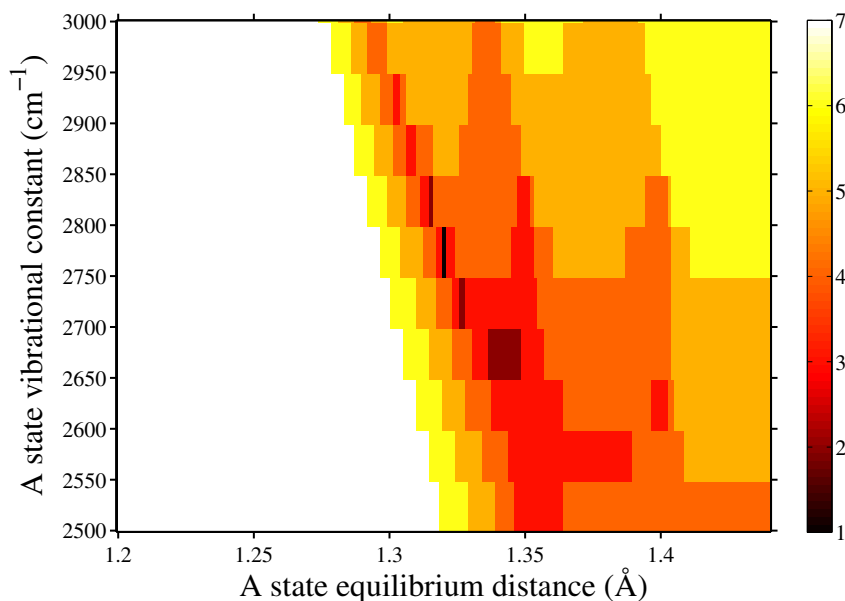


Figure 1. The effect of excited state vibrational constant and equilibrium bond length on the number of optically pumped vibrational states required in order to have less than one part-per-million population loss after spontaneous emission. Calculations are based on probabilistic analysis of FCFs of a hypothetical A–X system. White space represents parameter regions where at least seven vibrational bands must be addressed. The ground- and excited-state PECs are represented by Morse functions with a fixed anharmonicity of $\omega_e x_e$ of 50 cm^{-1} . The ground state, X, has an equilibrium distance R_e of 1.32 Å and a vibrational constant ω_e of 2750 cm^{-1} . Laser cooling with one laser is possible only in a narrow region (in black) where the A-state spectroscopic constants are similar enough to R_e and ω_e .

$K' = 1$ with negative parity to the excited state $v = 0$, $N = 0$, $K = 1$ with positive parity, where N is the rotational angular momentum of the molecule and K is the total angular momentum without spin. As a result, decay from the excited state is limited to those v allowed by FCFs and only via the Q branch. The desired transitions and the expected spontaneous emission channels are shown in figure 2.

This method requires the control of the initial angular momentum population, as not all the molecules in the experiment start out with the desired N or J . However, the preparation of a specific initial state of the molecular ion can be achieved by laser cooling the internal states [24–26], internal state cooling by interaction with cold neutral atoms [43], using a cryogenic ion trap [44] to limit the number of energetically accessible states and the rate at which they mix or using state-selective photoionization of a neutral molecule [23].

Additional rotational states can be populated by blackbody redistribution between different rotational states or by the decay of vibrationally excited states through spontaneous emission. This process does not occur for homonuclear diatomic ions. For heteronuclear diatomic ions, blackbody redistribution results in an increase in the number of required repumps if the redistribution occurs on cooling timescales. For BH^+ and AlH^+ we find that rotational

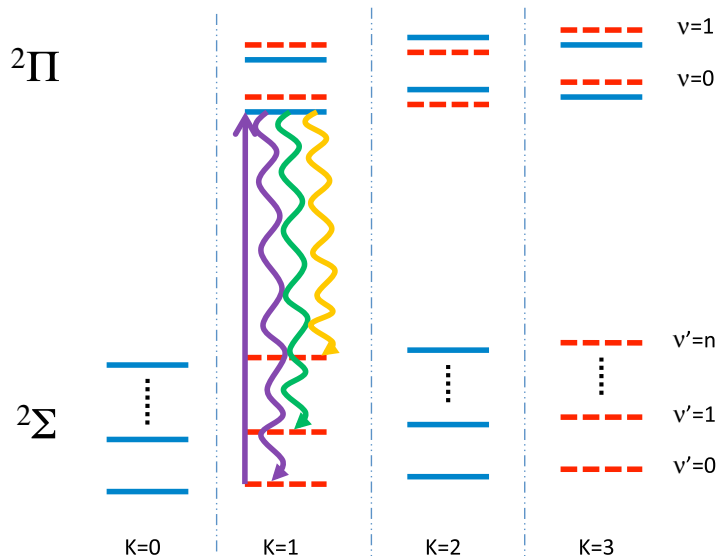


Figure 2. Energy level schematic of the proposed $^2\Pi-^2\Sigma$ transition. Red dashed levels are odd parity and blue solid levels are even parity. K refers to the total angular momentum without spin, and v and v' label the vibrational levels of the excited and ground states. Starting in the $K = 1$ state and exciting the Q -branch (solid arrows) results in a closed transition modulo the vibrational states (wiggly arrows).

redistribution requires tens of seconds, which is longer than the cooling time and is included in our simulation (see section 3.3). Vibrational radiative relaxation can limit the number of vibrational bands that must be addressed, at the cost of repumping from a larger set of rotational states due to the diffusion in J during the vibrational cascade.

If vibrational relaxation is sufficiently fast, a probabilistic FCF approach to determine repumping requirements is no longer valid, since the FCF approach assumes vibrational levels are infinitely long-lived. For example, if we cycle on the $v' = 0 \leftrightarrow v = 0$ transition and repump on the $v' = 1 \leftrightarrow v = 0$ transition, the ion will occasionally decay via the $v = 0 \rightarrow v' = 2 \rightarrow v' = 1$ and the $v' = 1 \rightarrow v' = 0$ channels, resulting in parity flips and the occupation of higher rotational states.

A simple figure of merit (FOM) for the probabilistic FCF approach is calculated by comparing the total time required to Doppler cool the molecule with the vibrational decay rate, since all levels are approximately equally populated at the steady state. The FOM is given by $\frac{A_{00'}}{N A_{1'0'}}$, in which $A_{00'}$ is the excited-state decay rate, N is the total number of photons scattered to Doppler cool and $A_{1'0'}$ is the vibrational decay rate. Using an estimate of $N \approx 10^5$, we calculate an FOM of 1, 20 and 250 for BH^+ , AlH^+ and SrF , respectively. We expect that the effect of vibrational decay is most significant for BH^+ , followed by AlH^+ , which have much smaller FOMs than does SrF . Our rate-equation simulations (see section 3.3) confirm this prediction, with vibrational decay in BH^+ and AlH^+ becoming important at short and long times, respectively, and vibrational decay in SrF unimportant on cooling timescales.

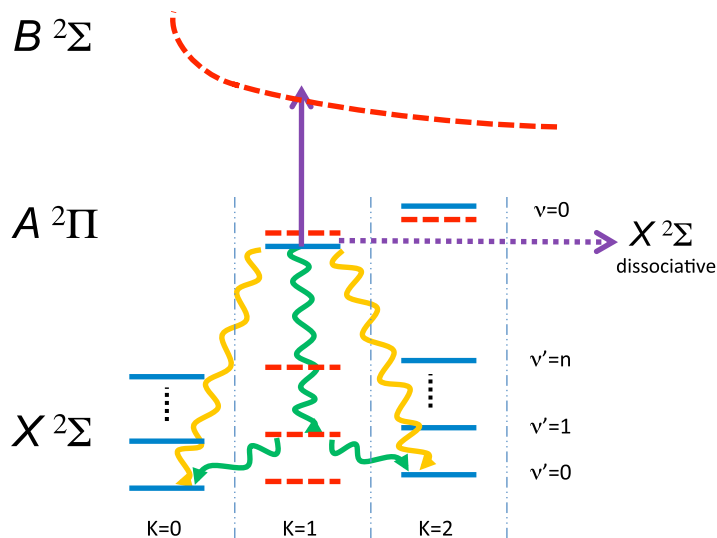


Figure 3. Energy level schematic of the proposed $^2\Pi-^2\Sigma$ transition showing transitions that result in either dissociation or occupation of a rotational state outside of the closed transition.

2.3. Other decay paths

The discussion so far has relied on the Born–Oppenheimer approximation and only considered dipole-allowed electronic transitions. These approximations reliably describe the molecules of interest for timescales up to a few microseconds. However, long ion-trap lifetimes require the consideration of slow processes driven by violations of the model.

Ideally, the A state is stable against dissociation. However, constant excitation of the molecule allows for rare events to occur, such as dissociation via the repulsive part of the ground state potential. This dissociation can be caused by spin–orbit coupling or L -uncoupling, which violates the Born–Oppenheimer approximation by mixing the ground and excited electronic state potentials [45]. In the case of L -uncoupling, it is possible to pick excited states such that symmetry effects forbid dissociation from occurring. Spin–orbit coupling also exists, but the calculated dissociation rate ranges from kHz to Hz and, to our knowledge, has never been measured. (In most molecular beam experiments these rare events are unobserved, since the dissociation timescales are long compared with laser-interaction timescales.) It is also possible that cooling or repumping lasers can cause photodissociation by coupling to the A state to a higher-lying dissociative state.

Additionally, transitions that are not electric dipole allowed can occur due to mixing of parity states by electric fields. For a compensated ion trap, the stray DC electric field at the axis can be effectively canceled [46]. Population of opposite parity states is inevitable in the long-time limit, due to magnetic dipole transitions (those of $\Delta J = 0, \pm 1$ terminate on states of the wrong parity). These transitions occur at a rate of 1 transition or less in 10^5 , also requiring attention when forming a highly closed cooling cycle [3].

Figure 3 graphically summarizes all the mechanisms by which the diatomic molecular ion can leave the cycling transition, including optical photon emission to an excited vibrational state, followed by the emission of an infrared photon that transfers the population to a lower vibrational state. As mentioned in the previous subsection, while this process leads to diffusion

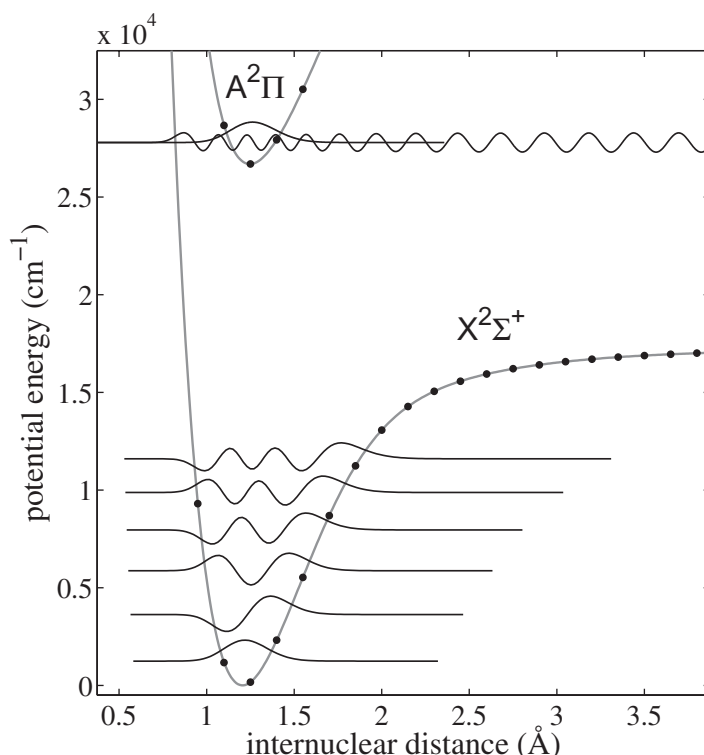


Figure 4. Born–Oppenheimer potential energy functions of the first two doublet electronic states of BH^+ at the FCI($3e^-$)/aug-cc-pV5Z level. Black dots are the actual FCI calculation results, and the solid lines are splines with analytical functions, as described in the LEVEL 8.0 and BCONT 2.2 manuals [64, 65]. The exact numerical solution of the radial Schrödinger equation yields the plotted wave functions, which are used to calculate the various decay and emission rates described in the text.

in the rotational states and introduces states of additional parity, it may ease the experimental requirements by reducing the number of vibrational bands that must be addressed.

3. Example laser cooling cycle for BH^+ and AlH^+

The system of both BH^+ and AlH^+ are well studied in the emission from hollow cathode discharges [30, 47, 48], and chemiluminescent ion–molecule reactions [35, 49] have yielded detailed rotational constants for the first two vibrational levels of the $\text{X}^2\Sigma^+$ ground state, along with low-resolution spectra of a few diagonal vibrational bands. For the BH^+ cation, 15 rovibrational energy levels of the ground state ($v = 0\text{--}4$ and $N = 0\text{--}2$) have been assigned by the extrapolation of photo-selected high-Rydberg series of neutral BH [32]. During those experiments, a very strong optical excitation of the ion core was observed when the energy of the scanning laser matched the A–X transition of the cation [50]. Photon absorption rates observed during this isolated core excitation were consistent with Einstein transition probability coefficients calculated theoretically by Klein *et al* [29]. Another favorable property of this molecular ion is that the first quartet states are energetically above the A–X system [31, 51],

eliminating the possibility of intervening electronic states of different spins to which the upper state could radiate and terminate the cycling transition. With all this information, it is possible to consider a laser cooling experiment for trapped BH^+ ions. Similarly, we expect AlH^+ to share most of these favorable characteristics.

3.1. Methods

3.1.1. Ab initio calculations for BH^+ and AlH^+ . To study the processes outlined in section 2, potential energy functions and dipole moments for the $\text{X}^2\Sigma^+$, $\text{A}^2\Pi$, $\text{B}^2\Sigma^+$ and $\text{a}^4\Pi$ states of BH^+ and AlH^+ are required. Additionally, the transition dipole moments and spin–orbit coupling matrix elements between these states are needed. For BH^+ , it is possible to solve the electronic Schrödinger equation exactly (within a basis) through full configuration interaction (FCI). By constraining the 1s orbital of B to be doubly occupied, a large aug-cc-pV5Z basis [52] can be used to obtain the potential energy functions. The spectroscopic constants obtained at this level of theory [FCI($3e^-$)/aug-cc-pV5Z] accurately reproduce the experimentally determined values (shown in table 1), and were computed by fitting a fourth-order polynomial to five energy points centered at R_e and evenly spaced by 0.005 Å. For AlH^+ , FCI was not applicable because correlating the important 2s and 2p electrons of Al leads to too many determinants in the expansion of the wave function. Coupled-cluster methods and their equation-of-motion variants provide a practical alternative [53, 54]. In particular, CC3 and EOM-CC3 [55] were applied with an aug-cc-pCVQZ basis set [52]. Again, the 1s orbital was constrained to be doubly occupied. The FCI and EOM-CC3 computations were performed with PSI 3.4 [56].

Dipole moments, transition dipole moments and spin–orbit coupling matrix elements were obtained from multireference CI wavefunctions (MRCI). The MRCI wavefunctions add single and double excitations to a state-averaged complete active space self-consistent field (SA-CASSCF) reference [57, 58]. The SA-CASSCF orbitals were optimized with an active space consisting of three electrons in the $2\sigma 1\pi_x 1\pi_y 3\sigma 4\sigma 5\sigma$ and $4\sigma 2\pi_x 2\pi_y 5\sigma 6\sigma 7\sigma$ orbitals for BH^+ and AlH^+ , respectively. In the MRCI wavefunctions, single and double excitations from the 1s B orbital and from the 2s and 2p Al orbitals were allowed. Again, the aug-cc-pV5Z and aug-cc-pCVQZ basis sets were used for BH^+ and AlH^+ , respectively. The MRCI computations were performed with MOLPRO [59].

All of the PECs, dipole moments, transition dipole moments and spin–orbit matrix elements (doublet–doublet, quartet–doublet and quartet–quartet) generated by the calculations described in this section have been included as a supplementary file (available from stacks.iop.org/NJP/13/063023/mmedia).

3.1.2. The Einstein \mathcal{A} and \mathcal{B} coefficients. The Einstein \mathcal{A} coefficients connecting two states are calculated using

$$\mathcal{A}_{u,l} = \frac{\omega_{u,l}^3}{3\hbar\pi c^3 \epsilon_0} \frac{g_{u,l}}{2J_u + 1} \mathcal{S}_{u,l} \mathcal{D}_{u,l}^2, \quad (1)$$

where $\omega_{u,l}/2\pi$ is the transition frequency, c is the speed of light, ϵ_0 is the permittivity of free space, $g_{u,l}$ is a degeneracy factor, J_u is the total angular momentum of the upper state, $\mathcal{S}_{u,l}$ is the Hönl–London factor, and $\mathcal{D}_{u,l}$ is the transition dipole-moment matrix element. The \mathcal{B} coefficient

Table 1. Spectroscopic constants for the $X^2\Sigma^+$ and $A^2\Pi$ states of BH^+ and AlH^+ . Previous theoretical values and the most precise experimental values, where available, are also shown for comparison. R_e is the equilibrium bond length, B_e , D_e and α_e parameterize the rotational energy level spacings [42], ω_e and $\omega_e x_e$ are the harmonic and anharmonic frequencies of the potential at equilibrium and D_0 is the dissociation energy. The units are cm^{-1} unless otherwise specified.

State	Method	Reference	R_e (Å)	B_e	D_e	α_e	ω_e	$\omega_e x_e$	D_0 (eV)
BH^+ $X^2\Sigma^+$	MRDCI	[28]	1.2059	12.552		0.542	2515.2	85.86	1.86
	MC-SCF	[29]	1.21	12.48		0.475	2492	64	1.78
	MC-SCF + CI	[31]	1.208	12.53		0.393	2594.8	74.94	1.92
	MC-SCF	[60]	1.2039	12.57	0.00135	0.426	2422.7	74.61	
	CCSDT(FC)-CBS/mix	[61]	1.2047	12.58			2524.7	64.4	1.99
	MRDCI	[62]	1.204	12.59			2548	74.8	1.97
$A^2\Pi$	MRCISD/aug-cc-pV5Z		1.20498	12.574	0.00125	0.47	2518.4	64.7	
	FCI($3e^-$)/aug-cc-pV5Z		1.20484	12.578	0.00125	0.47	2519.4	64.6	1.99
	Experiment	[30]	1.20292	12.6177	0.001225	0.4928	2526.8 ^a	61.98 ^a	1.95 ± 0.09^b
	MRDCI	[28]	1.2158	11.649		0.501	2228.3	66.88	3.33
	MC-SCF	[29]	1.253	11.62		0.467	2212	52	3.2
	MC-SCF + CI	[31]	1.247	11.76		0.41	2351.8	71.38	3.19
$B^2\Sigma^+$	MRDCI	[62]	1.247	11.75			2257	52	3.3
	MRCISD/aug-cc-pV5Z		1.24770	11.728	0.00128	0.46	2245.0	53.4	
	FCI($3e^-$)/aug-cc-pV5Z		1.24648	11.751	0.00128	0.46	2251.6	53.6	3.35
	Experiment	[30]	1.24397	11.7987		0.4543			
	MRDCI	[28]	1.9036	5.037		0.075	1263	33	1.28
	SDT-CI	[28]	1.9031	5.04		0.074	1258	31	1.28
$B^2\Sigma^+$	MC-SCF	[29]	1.912	4.99		0.097	1235	32	1.26
	MC-SCF + CI	[31]	1.91	5.01		0.073	1206	19	1.24
	MRDCI	[62]	1.889	5.12			1285	32	1.31
	MRCISD/aug-cc-pV5Z		1.90180	5.048	0.000322	0.074	1264.4	30.5	
	FCI($3e^-$)/aug-cc-pV5Z		1.90116	5.051	0.000323	0.075	1264.3	29.9	1.35

Table 1. Continued.

State	Method	Reference	$R_e(\text{\AA})$	B_e	D_e	α_e	ω_e	$\omega_e x_e$	$D_0(\text{eV})$
AlH^+ $X^2\Sigma^+$	MRDCI	[34]	1.6098	6.698		0.318	1684	81	0.666
	MC-SCF	[29]	1.608	6.71		0.317	1680	71	0.74
	MCQDPT	[37]	1.6	6.78		0.3945	1600	82	0.92
	MRCISD/aug-cc-pCVQZ		1.60802	6.711	0.000422	0.30	1692.4	66.2	
	CC3/aug-cc-pCVQZ		1.60543	6.732	0.000434	0.31	1677.0	69.1	0.73
	Experiment	[36]	1.605	6.736	0.0004469	0.382	1654	74	
$A^2\Pi$	MRDCI	[34]	1.6047	6.741		0.248	1727	54	1.75
	MC-SCF	[29]	1.609	6.7		0.251	1683	42	1.78
	MCQDPT ^c	[37]	1.6	6.7779		0.2668/0.2489	1703/1743	47/45	1.86/1.82
	MRCISD/aug-cc-pCVQZ		1.60375	6.746	0.000412	0.25	1726.3	33.2	
	EOM-CC3/aug-cc-pCVQZ		1.59126	6.853	0.000416	0.23	1759.3	42.8	1.90
	Experiment	[36]	1.595	6.817	0.0004152	0.243	1747	43	
$B^2\Sigma^+$	MRDCI	[34]	2.0582	4.097		0.046	1322	19	1.4
	MC-SCF	[29]	2.063	4.08		0.043	1326	21	1.4
	MRCISD/aug-cc-pCVQZ		2.05914	4.092	0.000155	0.039	1330.5	21	
	EOM-CC3/aug-cc-pCVQZ		2.03038	4.209	0.000158	0.039	1375.9	23	1.43

^a [32]; an $\omega_e y_e$ of $\approx -2 \text{ cm}^{-1}$ has been determined experimentally.^b [63].^c Values are for the $^2\Pi_{1/2}/^2\Pi_{3/2}$, respectively.

is calculated from the \mathcal{A} coefficient using

$$\mathcal{B}_{u,1} = \frac{\pi^2 c^3}{\hbar \omega_{u,1}^3} \mathcal{A}_{u,1}. \quad (2)$$

3.1.3. The rate-equation model. A rate-equation approach was used to model the population dynamics of BH^+ and AlH^+ , allowing the determination of the average number of Doppler-cooling photons scattered before a molecule is pumped into a dark state (ignoring predissociation and photodissociation described in section 3.3.3). A similar approach has been used in internal-state cooling [24–26]. In particular, we solved

$$\frac{d\mathbf{P}}{dt} = \mathbf{M}\mathbf{P}, \quad (3)$$

where \mathbf{P} is a vector consisting of the N rovibrational levels we include in our model, arranged in order of by increasing energy, and \mathbf{M} is an $N \times N$ vector consisting of the Einstein \mathcal{A} and \mathcal{B} coefficients which couple the different rovibrational states. Explicitly, the population of a given state follows

$$\begin{aligned} \frac{dP_i}{dt} = & - \sum_{j=1}^{j=i-1} \mathcal{A}_{ij} P_i - \sum_{j=1}^{j=i-1} \mathcal{B}_{ij} \rho(\omega_{ij}) P_i - \sum_{j=i+1}^{j=N} \mathcal{B}_{ij} \rho(\omega_{ij}) P_i \\ & + \sum_{j=i+1}^{j=N} \mathcal{A}_{ji} P_j + \sum_{j=1}^{j=i-1} \mathcal{B}_{ji} \rho(\omega_{ji}) P_j + \sum_{j=i+1}^{j=N} \mathcal{B}_{ji} \rho(\omega_{ji}) P_j. \end{aligned} \quad (4)$$

Here, the i th and j th states are connected by the Einstein coefficients denoted by \mathcal{A}_{ij} , \mathcal{B}_{ij} and \mathcal{B}_{ji} , which correspond to spontaneous emission, stimulated emission and absorption, respectively, and $\rho(\omega_{ij})$ is the spectral energy density at a given frequency, ω_{ij} . The Einstein coefficients were calculated using the PECs and dipole moments discussed in section 3.1.1.

The average number of Doppler cooling scattered photons was calculated by numerically solving equation (3) and multiplying the population in the excited state of the cycling transition by its spontaneous emission rate. Counting the photons emitted into the ground state from the excited state, regardless of whether the excited state was populated by a cycling-laser photon or a repumping-laser photon, is equivalent to counting the number of cooling-cycle absorption events. A simple estimate of the number of scattering events required for cooling can be obtained by considering the average energy lost per scattering event. For a laser detuning $\Gamma/2$ from resonance, where Γ is the natural linewidth, then the average energy lost per scatter is $\Delta E = \hbar\Gamma/2$. The linewidths of BH^+ and AlH^+ are $\Gamma = 2\pi \times 0.7$ MHz and $\Gamma = 2\pi \times 2.6$ MHz, respectively. Starting from $T = 300$ K, cooling to mK requires $N \approx 10^6$ scattering events for both BH^+ and AlH^+ for our laser detuning. We may also take advantage of the Doppler width at higher temperatures by detuning the laser further from resonance to increase the energy lost per scattering event, reducing the laser detuning as the ions are cooled. Using this method, we estimate that a lower limit of $N \approx 10^4$ scattering events are required to cool to mK.

Finally, it should be noted that a rate-equation approach does not account for coherence effects that can result in dark states. For example, in atomic systems with a low-lying D state, two lasers are typically used for Doppler cooling to address an $S \rightarrow P$ (cooling) transition and a $D \rightarrow P$ (repumping) transition. When both lasers are detuned from resonance by similar amounts, dark resonances occur and cycling terminates [66]. The cooling and repumping

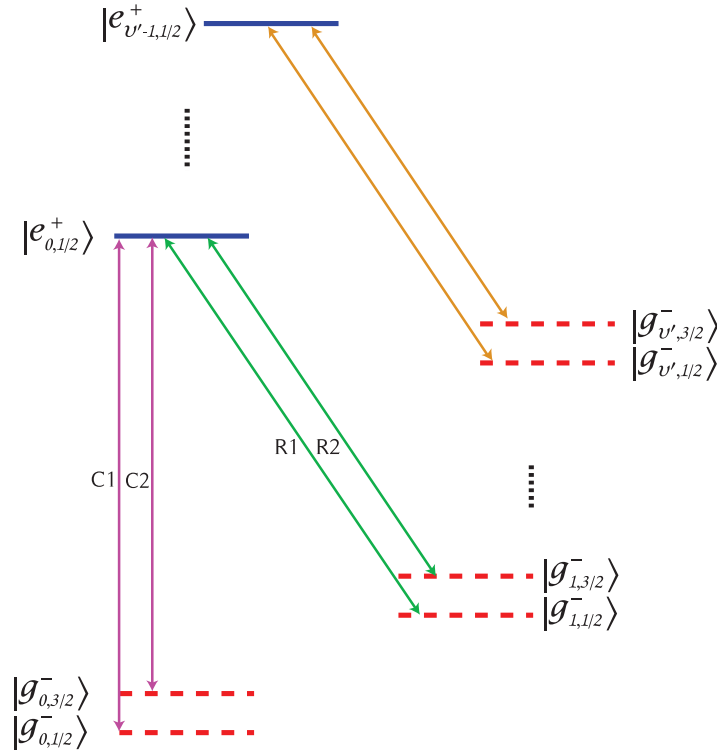


Figure 5. Energy level schematic diagram showing cooling and repumping transitions for the case of negligible vibrational and rotational relaxation within the excited- and ground-state manifolds (not to scale). Ground and excited states are labeled as $|g_{v',J'}^P>$ and $|e_{v,J}^P>$, respectively.

transitions in our molecular system also exhibit a similar Λ -system (see figures 5 and 6). Experimentally, these dark resonances can be avoided either by explicitly varying the laser detuning or by shifting the energy levels by different amounts with an externally applied time-varying field [66, 67].

3.2. Cooling scheme

The general cooling scheme, ignoring spin, was discussed in section 2 and figure 2. The inclusion of spin results in spin-rotation splitting for the $^2\Sigma^+$ states and spin-orbit splitting of the $^2\Pi$ states [42]. We label the $X^2\Sigma^+$ and $A^2\Pi_{1/2}$ states as $|g_{v',J'}^P>$ and $|e_{v,J}^P>$, respectively. The Zeeman substructure is explicitly ignored in our model; however, the resulting reduction in the overall scattering rate due to differing multiplicities of the ground and excited states is included as a degeneracy factor in the Einstein \mathcal{A} and \mathcal{B} coefficients. In practice, an externally applied magnetic field of a few Gauss is sufficient to lift the degeneracy, and the laser polarization can be chosen to ensure that all Zeeman sublevels are addressed [66, 67]. Additionally, both AlH^+ and BH^+ have hyperfine structure, which is ignored in our model.

We assume that population begins in the $|g_{0,1/2}^->$ state and a laser drives transitions to the $|e_{0,1/2}^+>$ state. From the excited state, population may decay into either the $|g_{v',1/2}^->$ state or the $|g_{v',3/2}^->$ state, with decreasing probability for increasing v' . The cooling transition,

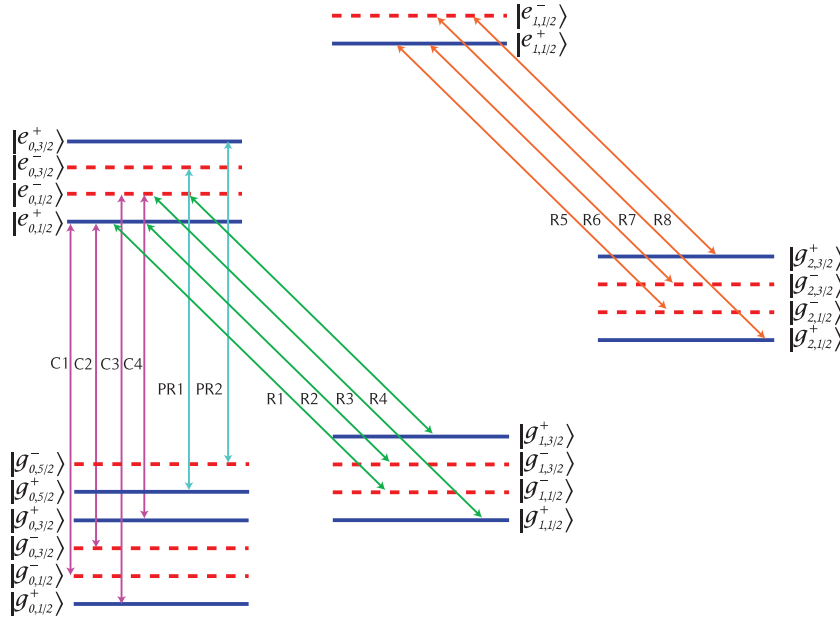


Figure 6. Energy level schematic diagram showing cooling and repumping transitions for the case of non-negligible vibrational and rotational transitions within the excited- and ground-state manifolds (not to scale).

corresponding to $v' = 0$, requires driving both the $|g_{0,1/2}^-\rangle \longleftrightarrow |e_{0,1/2}^+\rangle$ and $|g_{0,3/2}^-\rangle \longleftrightarrow |e_{0,1/2}^+\rangle$ transitions, as shown in figure 5 with labels C1 and C2. The ground states belong to the same rotational level ($K' = 1$), so this cooling scheme is equivalent to the one discussed in section 2.2.

If vibrational levels with $v' \geq 1$ are long-lived, then decay between vibrational levels, which result in parity flips, can be ignored. In this case, we drive $|g_{v',1/2}^-\rangle \longleftrightarrow |e_{v'-1,1/2}^+\rangle$ and $|g_{v',3/2}^-\rangle \longleftrightarrow |e_{v'-1,1/2}^+\rangle$ transitions to pump population back into the cycling transition, as shown by the R1, R2 and unlabeled arrows in figure 5.

However, when decay between vibrational levels is fast, the resulting parity flips may populate the $|g_{v',J'}^+\rangle$ states, so lasers connecting these states to an excited state must also be included to avoid population buildup in the even-parity states. Multi-stage vibrational decay also results in angular-momentum diffusion. For instance, decay from the $|e_{0,1/2}^+\rangle$ populates the $|g_{1,3/2}^-\rangle$ state, which decays into the $|g_{0,5/2}^+\rangle$ state. Population is pumped out of states with $J' > 3/2$ by driving $|g_{0,J'}^\pm\rangle \longleftrightarrow |e_{0,J'-1}^\mp\rangle$. This cooling scheme is illustrated in figure 6.

In figures 7 and 9, we show the rotationless lifetimes for BH^+ and AlH^+ , respectively. Since both AlH^+ and BH^+ have relatively high decay rates (up to 100 s^{-1}) between vibrational levels, we find that they require cycling and repumping on both parities. In the following section, we examine in detail each molecule separately, and determine their repumping requirements. Predissociation and photodissociation rates are considered in the last subsection.

3.3. Simulation results

3.3.1. BH^+ . In figure 8, the total number of scattered photons for different laser configurations is plotted. Population is initially in the $|g_{0,1/2}^-\rangle$ state. The simulation includes dipole-allowed transitions between electronic states, and vibrational and rotational transitions within the $X^2\Sigma^+$

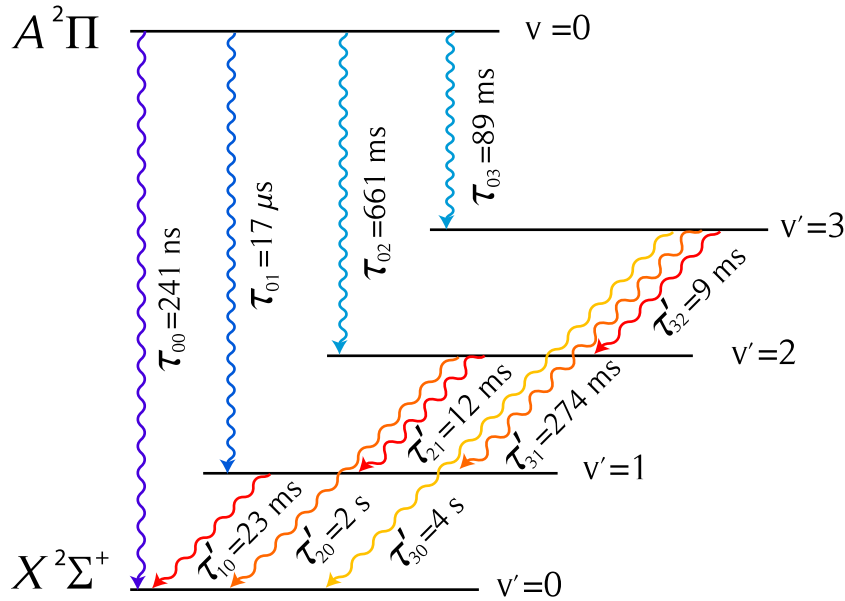


Figure 7. Energy level schematic diagram showing rotationless lifetimes of different BH^+ levels (not to scale). The lifetimes for $\text{A} \rightarrow \text{X}$ transitions are denoted by τ and the lifetimes for transitions within the ground state are denoted by τ' . All lifetimes are calculated using the *ab initio* PECs and dipole moments discussed in section 3.1.

state. Blackbody radiation is also included in the model, for a temperature of 300 K. In case (a-i), two wavelengths are used to drive the odd-parity cycling transitions: C1 and C2 in figures 5 and 6. Population is quickly pumped into the $v' = 1$ manifold, and a total of $N \approx 70$ photons are scattered. In case (a-ii), an additional two wavelengths are introduced to drive the even-parity cycling transitions: C3 and C4 (see figure 6). Again, population is quickly pumped into the $v' = 1$ manifold; however, at longer times ($t \approx \tau'_{10}$), population leaks back into the even-parity ground states and is driven by the even-parity cycling transitions, resulting in a total of $N \approx 380$ scattered photons. The delay between successive photon count plateaus is due to population building up in the $v' = 1$ manifold.

Case (b-i) includes the transitions from case (a-i), plus an additional two wavelengths to drive the odd-parity repumping transitions: R1 and R2 in figures 5 and 6. $N \approx 8000$ photons are scattered, with the dominant dark-state leak being the even-parity ground states. Case (b-ii) extends case (a-ii) with an additional four wavelengths to drive even- and odd-parity repumping transitions: R1–R4. A total of $N \approx 21\,000$ photons are scattered, which plateaus at just above the swept-detuning threshold.

As predicted in section 3.2, vibrational decay plays a significant role in depopulating the higher-lying vibrational ground states, resulting in parity flips and rotational diffusion. In case (b-ii), the parity-flip issue is addressed, so population built up in the $|g_{0,5/2}^{\pm}\rangle$ states. Case (c-ii) includes the same wavelengths as case (b-ii) and also two additional wavelengths to drive odd- and even-parity P-branch repumps: PR1, PR2 (see figure 6). The increase in counts ($N \approx 600\,000$), which is well above the swept-detuning threshold required for significant cooling from room temperature.

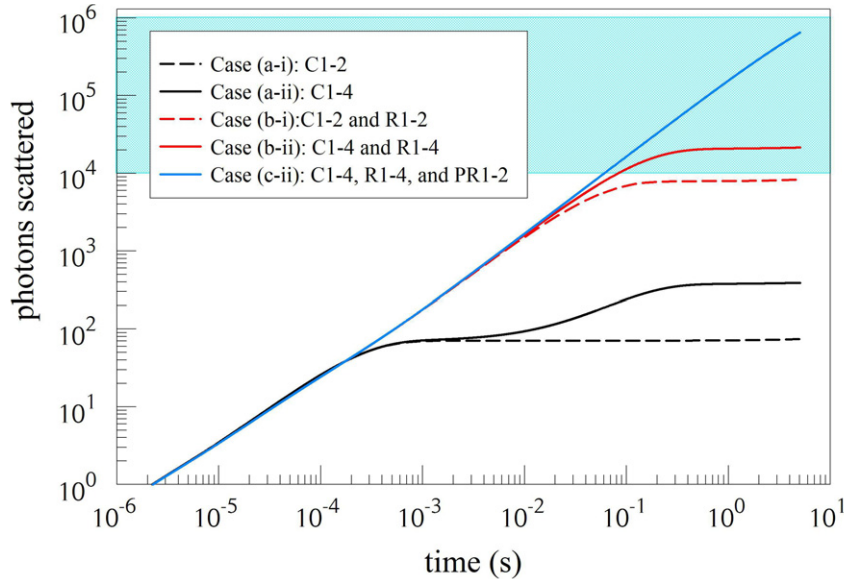


Figure 8. Plot of the integrated cycling-transition photon count for the BH^+ simulation using different laser configurations, driving the subsets of transitions shown in figure 6. The hatched region corresponds to the upper and lower limits for Doppler cooling to mK temperatures, corresponding to fixed and swept detuning, respectively.

3.3.2. AlH^+ . The simulation of AlH^+ is plotted in figure 10. Again, we assume that population starts in the $|g_{0,1/2}^- \rangle$ state. The simulations include transitions between electronic states, within the ground electronic state, and include a blackbody source of radiation at 300 K. In case (a), four wavelengths are used to drive only the cycling transition: C1–C4. A total of $N \approx 200$ photons are scattered.

In case (b) an additional four wavelengths are used to drive the repumping transitions: R1–R4. There is a significant increase in counts ($N \approx 15\,000$), and a noticeable deviation from the results of BH^+ . At $t \approx 10^{-3}$ s, there is a reduction in rate, resulting from population building up in the $v' = 2$ level. For BH^+ the rate of populating $v' = 2$ is 55 times *lower* than the decay from $v' = 2$ to $v' = 1$ ($\tau_{02}/\tau'_{21} = 55$). For AlH^+ the rate of populating $v' = 2$ is 500 times *higher* than the decay from $v' = 2$ to $v' = 1$ ($\tau_{02}/\tau'_{21} = 1/500$), resulting in population trapping.

In case (c) an additional four wavelengths are applied to drive the repumping transitions: R5–R8. This results in a total of $N \approx 210\,000$ photon counts, which is well above the swept-detuning cooling threshold. Interestingly, the second vibrational repump is sufficient for the cooling of AlH^+ , whereas BH^+ requires optical pumping of the $|g_{5/2}^\pm \rangle$ states.

In case (d) $\Delta J = -1$ repumping transitions are included: PR1 and PR2. The resultant photon count ($N \approx 700\,000$), which is similar to the number of counts we obtain in case (c-ii) of BH^+ . We find that the results of AlH^+ are similar to those of BH^+ in that they each require repumping of both parities, due to vibrational decay; however, AlH^+ differs in that it requires an additional vibrational repump while not requiring additional rotational repumping.

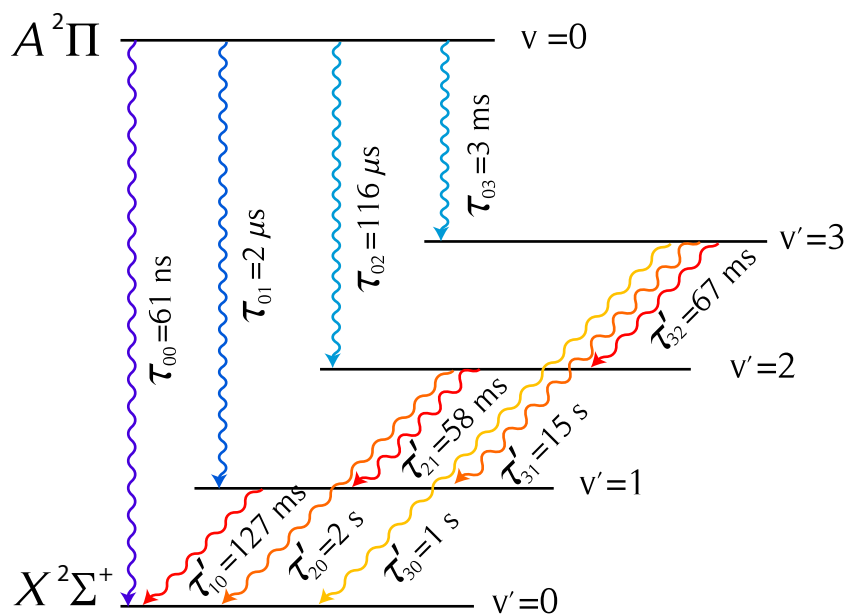


Figure 9. Energy level schematic showing rotationless lifetimes of different AlH^+ levels (not to scale). The lifetimes for $\text{X} \rightarrow \text{A}$ transitions are denoted by τ , and the lifetimes for transitions within the ground state are denoted by τ' . All lifetimes are calculated using the *ab initio* PECs and dipole moments discussed in section 3.1.

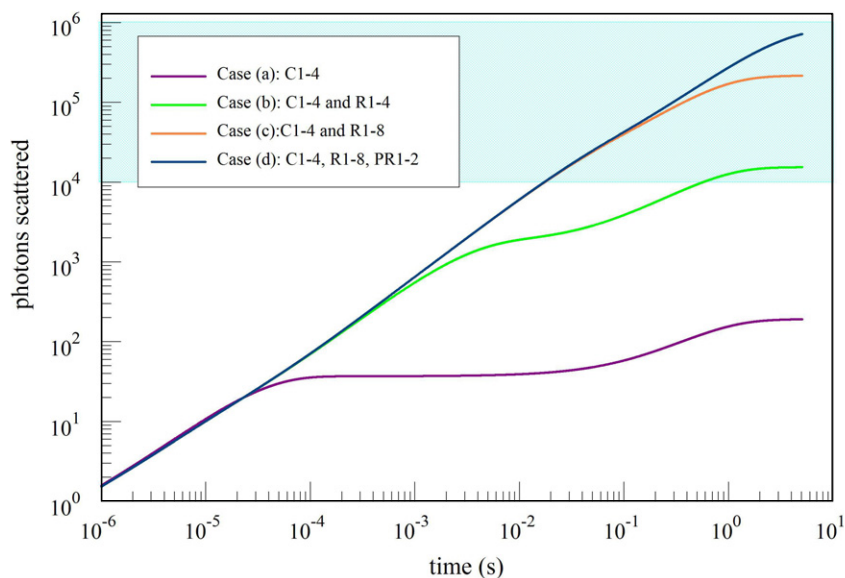


Figure 10. Plot of the integrated photon count for the AlH^+ simulation using different laser configurations, driving the subsets of transitions shown in figure 6. The hatched region corresponds to the upper and lower limits for Doppler cooling to mK temperatures, corresponding to fixed and swept detuning, respectively.

3.3.3. Calculation of predissociation and photofragmentation rates. The cooling scheme requires a continuous repopulation of the vibrational ground level of the $A^2\Pi$ excited electronic state. One possible mechanism that stops the cooling cycle is the isoenergetic predissociation of the molecular ion by the coupling of the bound excited state to a dissociative state of the $X^2\Sigma^+$ repulsive wall. Bound and continuum wavefunctions can interact via spin–orbit and rotational–orbit couplings. The expression for such predissociation rates, in units of s^{-1} , is obtained from Fermi’s golden rule [64]

$$k_s(v, J) = \frac{2\pi}{\hbar} |\langle \psi_{E,J'}(R) | M_s(R) | \psi_{v,J}(R) \rangle|^2, \quad (5)$$

where $M_s(R)$ is a term in the molecular Hamiltonian, which is neglected in the Born–Oppenheimer approximation and describes the coupling between the initial and final electronic states. The $|\psi_{v,J}(R)\rangle$ initial-state wavefunction is space-normalized and the $|\psi_{E,J'}(R)\rangle$ wavefunction of the continuum is energy-normalized [45]. The calculation uses a one-dimensional (1D) density of states in the bond coordinate to determine the energy normalization, $d(E) \propto \sqrt{\mu/(E-D)}$, where μ is the reduced mass and D is the dissociation energy of the final electronic state [64].

The predissociation depends strongly on the slope of the repulsive part of the dissociative potential. In order to estimate the uncertainty in the predissociation calculations, we first calculate the overlap integral of the wavefunctions assuming and multiplying by a bond-length-independent spin–orbit coupling of 13.9 cm^{-1} [30]. For BH^+ , by varying the details of the bound-state wavefunction, we find that the predissociation rate varies by more than four orders of magnitude from 10^2 to 10^{-2} s^{-1} . Representing the $A^2\Pi(v=0)$ wavefunction by a symmetric Gaussian from the solution of the harmonic oscillator, the rate results in the most disastrous scenario, $4.4 \times 10^2\text{ s}^{-1}$. Using the solution of a Morse potential gives $6.2 \times 10^{-2}\text{ s}^{-1}$. Combining the continuum and bound wave functions computed using LEVEL 8.0 and BCONT 2.2, respectively, and by inputting the FCI(3e[−])/aug-cc-pV5Z $A^2\Pi$ and $X^2\Sigma^+$ PECs, the predissociation rate is $1.6 \times 10^{-1}\text{ s}^{-1}$ (potentials and wavefunctions shown in figure 4). Including the spin-orbit coupling function calculated in section 3.1.1, as prescribed in equation (5), the predissociation rate results in a value of $0.03 \pm 0.02\text{ s}^{-1}$. For AlH^+ (using EOM-CC3 potential energy functions and MRCI spin–orbit coupling matrix elements), the predissociation rate is $0.2 \pm 0.1\text{ s}^{-1}$. The quoted uncertainties come from interpolation noise in both the repulsive part of the $X^2\Sigma^+$ state and the point piecewise polynomial used to fit the R-parametrized spin–orbit couplings. This uncertainty due to interpolation is negligible for the calculation of optical decay rates (~ 10 parts per million) reported elsewhere in this paper.

Another problematic event is the excitation from the $A^2\Pi$ state to either bound or dissociative $B^2\Sigma^+$ levels, by absorption of a cooling or repump photon. For BH^+ , PECs from two different *ab initio* levels of theory place the dissociation limit, V_{lim} , either below or above the energy accessible by the 379 and 417 nm photons excited from the A state (see table 2 and figure 11). Although both the results agree with predictions based on experimental observations (due to a 2000 cm^{-1} uncertainty), these calculations yield two different photon absorption scenarios. The first (MC-SCF) would photodissociate the molecule, and the second (FCI) would steal population from the cooling cycle to a bound vibrationally excited level of the $B^2\Sigma^+$ state. For AlH^+ , MC-SCF and EOM-CC3 predict V_{lim} to be below the total energy accessed by any of the three required cooling and repump wavelengths (360, 381 and 376 nm).

Table 2. Energies and rate coefficients relevant to possible photofragmentation events. $\hbar\omega$ (T), A (T) and I_s (T) correspond to the photon energy, the Einstein A coefficient and saturation intensity for each transition T, as labeled in figures 5 and 6. The BH^+ cooling schemes do not involve the transition R5. $M^+(\text{}^3\text{P})-M^+(\text{}^1\text{S})$ is the energy difference between the lowest singlet and triplet energy states of B^+ or Al^+ . This should correspond to the energy difference between the dissociative asymptotes of the $\text{X}^2\Sigma^+$ ground state and the common $\text{A}^2\Pi$ and $\text{B}^2\Sigma^+$ excited states V_{lim} . E_{total} is the energy of the $\text{A}^2\Pi$ ($v=0, J=1/2$) state plus the photon energy of the transition T. Photofragmentation does not occur in the FCI($3e^-$)/aug-cc-pV5Z) calculation for BH^+ (see text), but the $\text{X}^2\Sigma^+ \text{D}_e$ does not preclude either possibility due to current experimental uncertainties.

	BH^+			AlH^+		
	MC-SCF ^a	FCI	Observed ^b	MC-SCF ^a	EOM-CC3 /aug-cc-pCVQZ	Observed ^c
$\text{X}^2\Sigma^+(v=0, K=1)$ [cm^{-1}]	1223	1268	1272	810	848	
$\text{X}^2\Sigma^+(v=2, K=1)$ [cm^{-1}]				3669	3702	
$\hbar\omega$ (C1) [cm^{-1}]	27 594	26 547	26 356	28 521	27 747	27 673
$\hbar\omega$ (R1) [cm^{-1}]	25 239	24 163	23 960	27 001	26 225	
$\hbar\omega$ (R5) [cm^{-1}]				27 255	26 566	
\mathcal{A} (C1) [s^{-1}]	2.9×10^6	2.8×10^6		1.1×10^7	1.1×10^7	
\mathcal{A} (R1) [s^{-1}]	3.8×10^4	3.9×10^4		1.3×10^5	2.9×10^5	
\mathcal{A} (R5) [s^{-1}]				4.8×10^5	8.7×10^5	
I_s (C1) [photons/($\text{cm}^2 \text{s}$)]	2.3×10^{15}	2.0×10^{15}		9.6×10^{15}	8.8×10^{15}	
I_s (R1) [photons/($\text{cm}^2 \text{s}$)]	2.6×10^{13}	2.4×10^{13}		1.0×10^{14}	2.1×10^{14}	
I_s (R5) [photons/($\text{cm}^2 \text{s}$)]				3.7×10^{14}	6.4×10^{14}	
$\text{X}^2\Sigma^+ \text{D}_e$ [cm^{-1}]	15 645	17 261	$17\,000 \pm 2000$	5998	6694	≈ 6300
$M^+(\text{}^3\text{P})-M^+(\text{}^1\text{S})$ [cm^{-1}]	38 068	37 546	37 342	37 003	37 206	37 454
A and B V_{lim} [cm^{-1}]	53 713	54 807	$54\,450 \pm 2000$	43 000	43 900	$\approx 43\,750$
E_{Total} (C1) [cm^{-1}]	56 411	54 362	53 984	57 852	56 342	
E_{Total} (R1) [cm^{-1}]	54 056	51 978	51 588	56 332	54 820	
E_{Total} (R5) [cm^{-1}]				58 179	56 834	
σ_{abs} (C1) [cm^2]	1.1×10^{-20}			7.2×10^{-23}	3.5×10^{-22}	
σ_{abs} (R1) [cm^2]	7.5×10^{-21}			1.4×10^{-22}	7.1×10^{-22}	
σ_{abs} (R5) [cm^2]				5.4×10^{-22}	2.4×10^{-21}	
k_{diss} (C1) [s^{-1}]	2.6×10^{-5}			6.9×10^{-7}	3.1×10^{-6}	
k_{diss} (R1) [s^{-1}]	1.9×10^{-7}			1.4×10^{-8}	1.5×10^{-7}	
k_{diss} (R5) [s^{-1}]				2.0×10^{-7}	1.6×10^{-6}	

^a [29].

^b [30, 32, 63].

^c [47, 63].

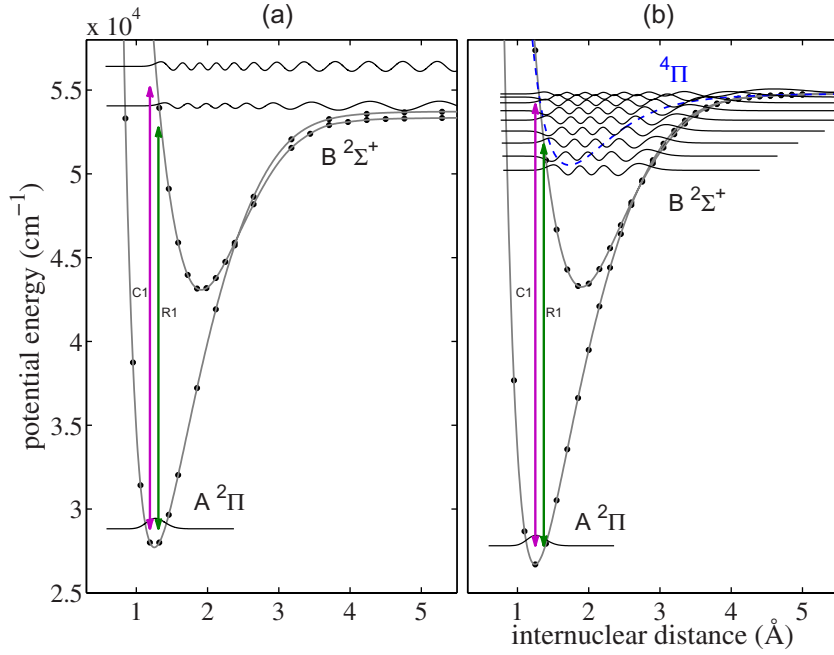


Figure 11. Sensitivity of problematic two-photon channels to precise details of the molecular structure. Potential energy functions of the $\Lambda = 1$ electronic states of BH^+ are plotted at the (a) MC-SCF level from [29] and (b) FCI(3e $^-$)/aug-cc-pV5Z. Black dots are the actual *ab initio* results and the solid lines are splines with analytical functions described in the BCONT 2.2 and LEVEL 8.0 manuals [64, 65]. The wavefunctions shown, which are used to calculate photofragmentation or photoabsorption rates, are positioned at the eigenvalues that solve the radial Schrödinger equation. The length of the arrows represents the actual experimental energy of the C1 and R1 photons.

The photodissociation cross-section (in cm^2 permolecule), is [68]

$$\sigma_s(\nu, J) = \frac{2\pi\omega}{3\hbar c} \sum_{J'} \frac{S_J^{J'}}{2J+1} |\langle \psi_{E,J'}(R) | M_s(R) | \psi_{\nu,J}(R) \rangle|^2. \quad (6)$$

S is the usual Hönl–London rotational intensity factor ($S_J^{J'}/(2J+1)$ is set to 1 in this section), and $M_s(R)$ is the transition moment function, including the ratio of initial to final-state electronic degeneracy factors. Allowed values of J' are given by the usual rotational selection rule. Similarly to equation (5), this expression assumes that the continuum radial wavefunction amplitude is energy normalized.

Assuming saturation intensities, calculated from the Einstein \mathcal{A} coefficients for the main cycling and repump transitions, we calculate the photon flux for each transition, I_s (T), listed in table 2. The photodissociation rates are the multiplication of these laser intensities by the cross-sections calculated from equation (6). Continuum and bound wavefunctions were calculated using BCONT 2.2 and LEVEL 8.0, respectively, and using the EOM-CC3 PECs and the MRCI transition moment functions. We perform simulations using the MC-SCF potential energies and transition dipole moments from [29] for comparison. We determine that photodissociation occurs on a timescale of tens of hours for BH^+ and several days for AlH^+ .

For BH^+ , the FCI PECs predict that the cycling and repump wavelengths result in an excitation from the $\text{A}^2\Pi$ state into a bound level of the $\text{B}^2\Sigma^+$ state. We believe that the probability of resonant absorption to the $\text{B}^2\Sigma^+$ state is small due to the narrow laser linewidths relative to the vibrational energy spacing and due to the poor Franck–Condon overlap. The most intense B – A band reaching this energy region of the $\text{B}^2\Sigma^+$ state is the (11, 0) with the value for \mathcal{A} of $1.6 \times 10^3 \text{ s}^{-1}$. Population of this state results in vibrational diffusion, which is not repumped in our cycling scheme.

Given the 2000 cm^{-1} experimental uncertainty in the dissociation energy, we do not have enough information to determine if the photoabsorption (predicted by FCI) or photodissociation (according to MC-SCF) processes will present a problem for Doppler cooling.

4. Prospects for cooling

Our simulations show that we require a total of eight wavelengths to Doppler cool BH^+ and AlH^+ (see figures 8 and 10); however, cooling does not necessarily require eight unique lasers. For both ions, two of the cycling transitions, C1 and C2, can be addressed by a single laser, since these two wavelengths differ only by the spin–rotation splitting ($\Delta\omega_{\text{sr}} \approx 540 \text{ MHz}$ for BH^+ [30] and $\Delta\omega_{\text{sr}} \approx 1.7 \text{ GHz}$ for AlH^+ [47]). This splitting can be achieved from a single laser source by using either an acousto-optic modulator (AOM) or an electro-optic modulator (EOM). The other two cycling transitions, C3 and C4, each require their own source since their splitting is large ($\Delta\omega \approx 1 \text{ THz}$ for BH^+ and $\Delta\omega \approx 2 \text{ THz}$ for AlH^+). The cycling transitions require CW lasers that are narrow relative to the electronic linewidth.

BH^+ requires driving the transitions R1–4, corresponding to repumping $v' = 1$. These transitions also span 1 THz, so they would require a total of three CW lasers, since spin–rotation splitting between R2 and R3 can be bridged by an AOM. However, repumping on all four transitions can be achieved with a single-femtosecond pulsed laser, with a typical bandwidth of 100 cm^{-1} (3 THz). Using femtosecond repumps generally requires optical pulse shaping, in order to avoid R-branch (rotational heating) transitions. Pulse-shaped pumping has been used to achieve vibrational cooling of Cs_2 molecules, with a resolution of 1 cm^{-1} (30 GHz) [69]—which is sufficient resolution for rotationally resolved repumping of hydrides. A femtosecond laser can also be used to drive the P-branch repumping transitions PR1 and PR2; however, these transitions would require a second laser since the R1–4 transitions differ from the PR1 and PR2 by more than 100 cm^{-1} (3 THz). It should be noted that repumping of PR1 and PR2 may not be necessary, since the total photon count exceeds the number required for Doppler cooling in the swept-detuning case (see figure 8).

AlH^+ requires driving transitions R1–4, corresponding to repumping $v' = 1$ and R5–8, corresponding to repumping $v' = 2$. The repump R1–4 span 2 THz, which is within the bandwidth of a femtosecond laser. Transitions R5–8 also span 2 THz; however, they do not overlap with transitions R1–4 ($\Delta\omega \approx 9 \text{ THz}$), so they would require a second femtosecond laser.

If a broadband source for repumping is used in conjunction with CW lasers for cooling, the laser requirements for BH^+ and AlH^+ are reduced to five and six lasers driving 10 and 14 different transitions, respectively. For BH^+ , repumping transitions R1–R4 require one femtosecond laser, and repumping transitions PR1–2 require a second femtosecond laser. For AlH^+ , repumping transitions R1–R4 require one femtosecond laser, and repumping transitions R5–R8 require a second femtosecond laser. The PR1–2 wavelengths would require a third femtosecond laser, but they are not strictly necessary (see figure 10).

The comb-like nature of the frequency components of a femtosecond laser can be problematic if a transition which we wish to drive happens to fall between two comb lines (typically separated by 80MHz). Various modulation techniques, either internal or external to the femtosecond laser, can be used to address this complication.

5. Conclusions

The key element for direct laser cooling of molecules [1–4] and molecular ions are diagonal FCFs. From our survey of a number of molecules (see the [appendix](#)) we determined BH^+ and AlH^+ to be favorable candidates since a probabilistic FCF analysis (excluding vibrational decay) suggested that three and two vibrational bands, respectively, needed to be addressed to scatter 10^4 photons. A detailed analysis of state population dynamics (including vibrational decay) has shown that BH^+ requires addressing two vibrational bands while AlH^+ requires addressing three. For both molecular ions we find that vibrational decays result in diffusion of both parity and angular momentum; thus, implementation of this cooling scheme requires addressing the molecules with a large number of repump wavelengths.

To minimize the total number of lasers required, a femtosecond laser can be used, since multiple transitions fall within the laser bandwidth. However, for many of these transitions, the frequency of excitation needs to first be experimentally determined with higher accuracy. The spectroscopy can be performed and the direct cooling technique optimized by initially trapping the molecular ions with laser-cooled atomic ions and monitoring the sympathetic heating and cooling of the atomic ion by the molecular ion [70]. Ultimately, dissociation of the molecular ion represents a fundamental limit for the continuous Doppler cooling of the species presented here.

The attempted laser cooling of BH^+ and AlH^+ will result in the first directly laser-cooled molecular ions and/or accurate measurement of rare predissociation events. Either way, it will improve our knowledge of the spectroscopic splittings of BH^+ and AlH^+ by an order of magnitude. Together with experimental dissociation rates, measurements of these splittings yield the most stringent test of advanced electronic structure theories; as the five-valence-electron open-shell diatomic BH^+ is among the simplest stable molecules and serves as a prototype for the development and testing of such theories [28, 29, 31, 51], [61–63], [71–73].

More generally, any molecular ion can be sympathetically cooled, and the long trapping lifetime allows slow processes to be investigated. Thus, sympathetically cooled molecular ions could serve as a valuable tool in determining the rates of higher-order processes such as predissociation and photofragmentation, which are of concern for direct Doppler cooling applications.

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Appendix. Potential candidates

Doppler cooling candidates are divided into two main categories. The first category corresponds to transitions in which the electron is excited from the HOMO to the LUMO, and the second category corresponds to transitions in which a hole moves from one orbital to another. Each category is subdivided according to the grouping on the periodic table. Groups that we discuss are chosen, based on chemical intuition, to form open-shells with vertical excitations that do not perturb the chemical bond. Probabilistic predictions in this [appendix](#) are based on FCFs obtained from Morse potentials defined by spectroscopic constants available in the literature. Detailed population-dynamics studies are needed for a more rigorous selection.

A.1. Diatomic molecular ions with transitions that excite a single electron from the HOMO to the LUMO

A.1.1. Group 3 hydrides and halides. ScH^+ has an interesting $^2\Delta - ^2\Pi$ system with very similar spectroscopic constants for both electronic states; however, the transition is deep into the IR ($\lambda = 6 \mu\text{m}$) [74]. The ground state of YH^+ is a $^2\Sigma$ and the first excited state is a $^2\Delta$. Spectroscopic constants r_e and ω_e are similar but the transitions fall in the IR at $\approx 3.3 \mu\text{m}$ [75]. Finally, LaH^+ has a $^2\Sigma^+ - ^2\Delta$ system with similar spectroscopic constants and the transition is at $\approx 5.1 \mu\text{m}$ [76].

A.1.2. Group 13 hydrides and halides. The open-shells from Group 13 and halogens are stable with $^2\Sigma$ ground states. Unfortunately, the $^2\Pi$ potential curve is dissociative. This is common in the following examples: AlCl^+ , AlF^+ [77, 78] and GaCl^+ . Although GaF^+ is believed to have strong ionic character [79] it also has a dissociative first excited state [80]. The hydrides BH^+ and AlH^+ seem to be suitable for direct laser cooling and are the subject of the present paper. Experimental spectroscopic data for GaH^+ are not yet available [81].

A.1.3. Bialkalis, alkali monohalides, alkali monohydrides and molecular ions formed by alkalis and Group 13. Li_2^+ and LiNa^+ have $^2\Sigma_g^+$ and $^2\Sigma^+$ ground electronic states with only one valence electron, and they do not have valence correlation energy [82]. Their ground state dissociation energies are relatively high ($10\,000 \text{ cm}^{-1}$) and they do not seem to have dissociative states that couple non-adiabatically to the system. The lack of spectroscopic data on these systems stop us from making further predictions, but it would not be surprising if a few of these molecular ions have diagonal FCFs. These systems cannot be ruled out as potential candidates. The same seems to be true for LiAl^+ with a bound ground state and a dissociation energy of roughly 1 eV [82]. LiB^+ has a weaker bond and the A and X state spectroscopic constants are different [83].

With respect to the alkali monohydrides, although they are technically open-shells with one unpaired electron spin, they have very weak bonds. This is expected as the same unpaired electron shares some density to participate in the bond (e.g. KH^+ [84] and NaH^+ [85]).

A.1.4. Group 15 halides and hydrides. PH^+ and PF^+ would require repumping of six and ten vibrational bands to scatter 2×10^4 and 9×10^4 photons, respectively. They do not have nuclear spin and their ground states are $^2\Pi$. The first excited state of PH^+ is a $^2\Delta$ and the vertical transition from the ground state is at 381 nm. For the PF^+ case, the $\text{A}^2\Sigma - \text{X}^2\Pi$ optical transition is in the UV at 282 nm [86].

A.2. Diatomic molecular ions with an unpaired electron in which an optical transition moves a hole to a bonding orbital

A.2.1. Group 17 hydrides. Using Morse potentials derived from spectroscopic constants compiled by Radzigs and Smirnov [86], we notice the $A^2\Sigma-X^2\Pi$ transitions in molecules such as HF^+ , HCl^+ and HBr^+ to have very un-diagonal FCFs. However, a recent theoretical proposal to measure molecular parity violation [40] points out that HI^+ could have highly diagonal FCFs. This is attributed to the center of mass being very close to the iodine atom, resulting in the molecular orbit having mostly an atomic character. There is not enough spectroscopic information on excited states of HI^+ to design a laser cooling experiment [87].

A.2.2. Group 15 diatomic ions. We use FCFs from the spectroscopic constants compiled by Radzigs and Smirnov [86] to test N_2^+ and P_2^+ . The former could scatter 11×10^4 photons if irradiated by five lasers. These transitions fall in the near IR with the most energetic transition at 1091 nm. This homonuclear molecular ion has recently been loaded and sympathetically cooled in its rovibrational ground state [23]. Although P_2^+ has a $^2\Pi_u$ ground state with similar r_e and ω_e constants as its first $^2\Sigma_g^+$ excited state, the FCFs show that one would have to address more than eight transitions with $\lambda > 4.6 \mu\text{m}$ to scatter over 10^4 photons.

A.2.3. Group 5 and 6 halides. These molecules have crowded excited state manifolds and the first electronic states high multiplicity. VF^+ has an $A^4\Delta-X^4\Pi$ system with very similar r_e and ω_e constants and a vertical transition at $\approx 7 \mu\text{m}$. If it is possible to cope with higher multiplicity ($S = 5$), CrF^+ might work as it has an $A^5\Pi-X^5\Sigma^+$ with an IR transition ($\lambda = 2 \mu\text{m}$) [88].

References

- [1] Shuman E S, Barry J F, Glenn D R and DeMille D 2009 Radiative force from optical cycling on a diatomic molecule *Phys. Rev. Lett.* **103** 223001
- [2] Shuman E S, Barry J F and DeMille D 2010 Laser cooling of a diatomic molecule *Nature* **467** 820–3
- [3] Di Rosa M D 2004 Laser-cooling molecules—Concept, candidates and supporting hyperfine-resolved measurements of rotational lines in the A–X(0,0) band of CaH *Eur. Phys. J. D* **31** 395–402
- [4] Stuhl B K, Sawyer B C, Wang D and Ye J 2008 Magneto-optical trap for polar molecules *Phys. Rev. Lett.* **101** 243002
- [5] Drewsen M, Jensen I, Lindballe J, Nissen N, Martinussen R, Mortensen A, Staunum P and Voigt D 2003 Ion Coulomb crystals: a tool for studying ion processes *Int. J. Mass Spectrom.* **229** 83–91
- [6] Ryjkov V L, Zhao X and Schuessler H A 2006 Sympathetic cooling of fullerene ions by laser-cooled Mg^+ ions in a linear rf trap *Phys. Rev. A* **74** 023401
- [7] Willitsch S, Bell M T, Gingell A D and Softley T P 2008 Chemical applications of laser- and sympathetically-cooled ions in ion traps *Phys. Chem. Chem. Phys.* **10** 7200–10
- [8] Baba T and Waki I 1996 Cooling and mass-analysis of molecules using laser-cooled atoms *Japan. J. Appl. Phys.* **35** L1134–7
- [9] Drewsen M, Mortensen A, Martinussen R, Staunum P and Sorensen J L 2004 Nondestructive identification of cold and extremely localized single molecular ions *Phys. Rev. Lett.* **93** 243201
- [10] Koelemeij J C J, Roth B, Wicht A, Ernsting I and Schiller S 2007 Vibrational spectroscopy of HD^+ with 2-ppb accuracy *Phys. Rev. Lett.* **98** 173002
- [11] Rosenband T *et al* Observation of the $^1S_0 \rightarrow ^3P_0$ clock transition in $^{27}\text{Al}^+$ *Phys. Rev. Lett.* **98** 220801

- [12] Wolf A L, van den Berg S A, Gohle C, Salumbides E J, Ubachs W and Eikema K S E 2008 Frequency metrology on the $4s^2S_{1/2}-4p^2P_{1/2}$ transition in $^{40}\text{Ca}^+$ for a comparison with quasar data *Phys. Rev. A* **78** 032511
- [13] Herrmann M, Batteiger V, Knünz S, Saathoff G and Udem Th Hänsch T W 2009 Frequency metrology on single trapped ions in the weak binding limit: the $3s_{1/2}-3p_{3/2}$ transition in $^{24}\text{Mg}^+$ *Phys. Rev. Lett.* **102** 013006
- [14] Batteiger V, Knünz S, Herrmann M, Saathoff G, Schüssler H A, Bernhardt B, Wilken T, Holzwarth R, Hänsch T W and Udem Th 2009 Precision spectroscopy of the $3s-3p$ fine-structure doublet in Mg^+ *Phys. Rev. A* **80** 022503
- [15] Wolf A L, van den Berg S A, Ubachs W and Eikema K S E 2009 Direct frequency comb spectroscopy of trapped ions *Phys. Rev. Lett.* **102** 223901
- [16] Højbjerg K, Offenberger D, Bisgaard C Z, Stapelfeldt H, Staunum P F, Mortensen A and Drewsen M 2008 Consecutive photodissociation of a single complex molecular ion *Phys. Rev. A* **77** 030702
- [17] Offenberger D, Wellers C, Zhang C B, Roth B and Schiller S 2009 Measurement of small photodestruction rates of cold, charged biomolecules in an ion trap *J. Phys. B: At. Mol. Opt. Phys.* **42** 035101
- [18] Roth B, Blythe P, Wenz H, Daerr H and Schiller S 2006 Ion-neutral chemical reactions between ultracold localized ions and neutral molecules with single-particle resolution *Phys. Rev. A* **73** 042712
- [19] Okada K, Wada M, Boesten L, Nakamura T, Katayama I and Ohtani S 2003 Acceleration of the chemical reaction of trapped Ca^+ ions with H_2O molecules by laser excitation *J. Phys. B: At. Mol. Opt. Phys.* **36** 33–46
- [20] Staunum P F, Højbjerg K, Wester R and Drewsen M 2008 Probing isotope effects in chemical reactions using single ions *Phys. Rev. Lett.* **100** 243003
- [21] Willitsch S, Bell M T, Gingell A D, Procter S R and Softley T P 2008 Cold reactive collisions between laser-cooled ions and velocity-selected neutral molecules *Phys. Rev. Lett.* **100** 043203
- [22] Gingell A D, Bell M T, Oldham J M, Softley T P and Harvey J N 2010 Cold chemistry with electronically excited Ca^+ Coulomb crystals *J. Chem. Phys.* **133** 194302–13
- [23] Tong X, Winney A H and Willitsch S 2010 Sympathetic cooling of molecular ions in selected rotational and vibrational states produced by threshold photoionization *Phys. Rev. Lett.* **105** 143001
- [24] Vogelius I S, Madsen L B and Drewsen M 2004 Rotational cooling of heteronuclear molecular ions with $^1\Sigma$, $^2\Sigma$, $^3\Sigma$ and $^2\Pi$ electronic ground states *Phys. Rev. A* **70** 053412
- [25] Staunum P F, Højbjerg K, Skyt P S, Hansen A K and Drewsen M 2010 Rotational laser cooling of vibrationally and translationally cold molecular ions *Nat. Phys.* **6** 271–4
- [26] Schneider T, Roth B, Duncker H, Ernsting I and Schiller S 2010 All-optical preparation of molecular ions in the rovibrational ground state *Nat. Phys.* **6** 275–8
- [27] Schuster D I, Lev Bishop S, Chuang I L, DeMille D and Schoelkopf R J 2011 Cavity QED in a molecular ion trap *Phys. Rev. A* **83** 012311
- [28] Guest M F and Hirst D M 1981 The potential-energy curves of BH^+ *Chem. Phys. Lett.* **80** 131–4
- [29] Klein R, Rosmus P and Werner H J 1982 *Ab initio* calculations of low-lying states of the BH^+ and AlH^+ ions *J. Chem. Phys.* **77** 3559–70
- [30] Ramsay D A and Sarre P J 1982 High-resolution study of the $A^2\Pi-X^2\Sigma^+$ band system of BH^+ *J. Chem. Soc., Faraday Trans. 2* **78** 1331–8
- [31] Kusunoki I 1984 *Ab initio* calculations of the doublet and quartet states of BH^+ *Chem. Phys. Lett.* **105** 175–9
- [32] Viteri C R, Gilkison A T, Rixon S J and Grant E R 2006 Rovibrational characterization of $X^2\Sigma^+{}^{11}\text{BH}^+$ by the extrapolation of photoselected high Rydberg series in ^{11}BH *J. Chem. Phys.* **124** 144312
- [33] Shi D-H, Liu H, Zhang J-P, Sun J-F, Liu Y-F and Zhu Z-L 2010 Spectroscopic investigations on $\text{BH}^+(X^2\Sigma^+)$ ion using MRCI method and correlation-consistent sextuple basis set augmented with diffuse functions *Int. J. Quantum Chem.* **111** 2171–9
- [34] Guest M F and Hirst D M 1981 Potential-energy curves for the ground and excited-states of AlH^+ *Chem. Phys. Lett.* **84** 167–71

- [35] Muller B and Ottinger C 1986 Chemiluminescent reactions of second-row atomic ions. I. $\text{Al}^+ + \text{H}_2 \rightarrow \text{AlH}^+(\text{A}^2\Pi, \text{B}^2\Sigma^+) + \text{H}$ *J. Chem. Phys.* **85** 232–42
- [36] Muller B and Ottinger C 1988 The spectroscopic constants of the $\text{A}^2\Pi$ - and $\text{X}^2\Sigma$ -states of AlH^+ *Z. Naturforsch. A* **43** 1007–8
- [37] Li G X, Gao T and Zhang Y G 2008 The splitting of low-lying or low excited states for hydride molecules (cations) of the third period under spin–orbit coupling *Chin. Phys. B* **17** 2040–7
- [38] Wineland D J and Wayne Itano M 1979 Laser cooling of atoms *Phys. Rev. A* **20** 1521–40
- [39] Metcalf H J and van der Straten P 1999 *Laser Cooling and Trapping* (New York: Springer)
- [40] Isaev T A, Hoekstra S and Berger R 2010 Laser-cooled RaF as a promising candidate to measure molecular parity violation *Phys. Rev. A* **82** 052521
- [41] Miller J 2010 Optical cycling paves the way for laser-cooled molecules *Phys. Today* **63** 9–10
- [42] Herzberg G 1950 Molecular spectra and molecular structure. I. Spectra of diatomic molecules 2nd edn *Molecular Spectra and Molecular Structure* (New York: Van Nostrand-Reinhold)
- [43] Hudson E R 2009 Method for producing ultracold molecular ions *Phys. Rev. A* **79** 032716
- [44] Labaziewicz J, Ge Y, Antohi P, Leibbrandt D R, Brown K R and Chuang I L 2008 Suppression of heating rates in cryogenic surface-electrode ion traps *Phys. Rev. Lett.* **100** 013001
- [45] Field R W and Lefebvre-Brion H 2004 *The Spectra and Dynamics of Diatomic Molecules* (San Diego: Elsevier)
- [46] Berkeland D J, Miller J D, Bergquist J C, Itano W M and Wineland D J 1998 Minimization of ion micromotion in Paul trap *J. Appl. Phys.* **83** 5025–33
- [47] Almy G M and Watson M C 1934 The band spectrum of ionized aluminum hydride *Phys. Rev.* **45** 0871–6
- [48] Almy G M and Horsfall R B 1937 The spectra of neutral and ionized boron hydride *Phys. Rev.* **51** 491–500
- [49] Ottinger C and Reichmuth J 1981 Chemiluminescent ion–molecule reactions $\text{B}^+ + \text{H}_2$ *J. Chem. Phys.* **74** 928–33
- [50] Viteri C R, Gilkison A T, Rixon S J and Grant E R 2007 Isolated core excitation of ^{11}BH : photoabsorption in competition with Rydberg predissociation *Phys. Rev. A* **75** 013410
- [51] Hirata S, Nooijen M and Bartlett R J 2000 High-order determinantal equation-of-motion coupled-cluster calculations for ionized and electron-attached states *Chem. Phys. Lett.* **328** 459–68
- [52] Kendall R A, Dunning T H and Harrison R J 1992 Electron affinities of the first-row atoms revisited. Systematic basis sets and wave functions *J. Chem. Phys.* **96** 6796
- [53] Purvis G D and Bartlett R J 1982 A full coupled-cluster singles and doubles model: the inclusion of disconnected triples *J. Chem. Phys.* **76** 1910
- [54] Stanton J F and Bartlett R J 1993 The equation of motion coupled-cluster method. A systematic biorthogonal approach to molecular excitation energies, transition probabilities, and excited state properties *J. Chem. Phys.* **98** 7029
- [55] Koch H, Christiansen O, Jørgensen P and Olsen J 1995 Excitation energies of BH , CH_2 , and Ne in full configuration interaction and the hierarchy CCS , CC2 , CCSD , and CC3 of coupled cluster models *Chem. Phys. Lett.* **244** 75
- [56] Crawford T D *et al* PSI3: an open-source *ab initio* electronic structure package *J. Comput. Chem.* **28** 1610
- [57] Ruedenberg K, Cheung L M and Elbert S T 1979 MCSCF optimization through combined use of natural orbitals and the Brillouin–Levy–Berthier theorem *Int. J. Quantum Chem.* **16** 1069
- [58] Roos B O, Taylor P R and Siegbahn P E M 1980 A complete active space SCF method (CASSCF) using a density matrix formulated super-CI approach *Chem. Phys.* **48** 157
- [59] Werner H-J *et al* 2009 *MOLPRO, Version 2009.1, a Package of Ab Initio programs* <http://www.molpro.net>
- [60] Biskupic S and Klein R 1988 MC SCF study of potential curves of small radicals *Theochem—J. Mol. Struct.* **47** 27–31
- [61] Feller D and Sordo J A 2000 A CCSDT study of the effects of higher order correlation on spectroscopic constants. I. First row diatomic hydrides *J. Chem. Phys.* **112** 5604–10

- [62] Petsalakis I D and Theodorakopoulos G 2006 Multireference configuration interaction and quantum defect calculations on the Rydberg states of the BH molecule *Mol. Phys.* **104** 103–13
- [63] Rosmus P and Meyer W 1977 PNO-CI and CEPA studies of electron correlation effects. IV. Ionization energies of the first and second row diatomic hydrides and the spectroscopic constants of their ions *J. Chem. Phys.* **66** 13–9
- [64] Le Roy R J and Kraemer G T 2004 BCONT 2.2: computer program for calculating absorption coefficients, emission intensities or (golden rule) predissociation rates *Chemical Physics Research Report CP-650R²* University of Waterloo (the source code and manual for this program may be obtained from <http://leroy.uwaterloo.ca/programs/>)
- [65] Le Roy R J 2007 LEVEL 8.0: a computer program for solving the radial Schrödinger equation for bound and quasibound levels *Chemical Physics Research Report CP-663* University of Waterloo. (the source code and manual for this program may be obtained from <http://leroy.uwaterloo.ca/programs/>)
- [66] Janik G, Nagourney W and Dehmelt H 1985 Doppler-free optical spectroscopy on the Ba⁺ mono-ion oscillator *J. Opt. Soc. Am. B* **2** 1251–7
- [67] Berkeland D J and Boshier M G 2002 Destabilization of dark states and optical spectroscopy in Zeeman-degenerate atomic systems *Phys. Rev. A* **65** 033413
- [68] Le Roy R J 1976 Macdonald R G and Burns G. Diatom potential curves and transition moment functions from continuum absorption coefficients: Br₂ *J. Chem. Phys.* **65** 1485–500
- [69] Viteau M, Chotia A, Allegrini M, Bouloufa N, Dulieu O, Comparat D and Pillet P 2008 Optical pumping and vibrational cooling of molecules *Science* **321** 232–4
- [70] Clark C R, Goeters J E, Dodia Y K, Viteri C R and Brown K R 2010 Detection of single-ion spectra by Coulomb-crystal heating *Phys. Rev. A* **81** 043428
- [71] Cooper D L, Gerratt J and Raimondi M 1986 Potential-energy surfaces for the reaction of B+(S-1,P-3) with H₂ using spin-coupled VB theory—asymptotic regions of the surfaces *Chem. Phys. Lett.* **127** 600–8
- [72] Curtiss L A and Pople J A 1988 A theoretical study of the energies of BH_n compounds *J. Chem. Phys.* **89** 614–5
- [73] Ishida M, Toyota K, Ehara M and Nakatsuji H 2001 Analytical energy gradient of high-spin multiplet state calculated by the SAC-CI method *Chem. Phys. Lett.* **350** 351–8
- [74] Alvarado-Swaigood A E and Harrison J F 1985 Electronic and geometric structures of scandium hydride cations (ScH⁺ and ScH²⁺) *J. Chem. Phys.* **89** 5198–202
- [75] Pettersson L G M, Bauschlicher C W, Langhoff S R and Partridge H 1987 Positive ions of the first- and second-row transition-metal hydrides *J. Chem. Phys.* **87** 481–92
- [76] Das K K and Balasubramanian K 1991 Potential-energy surfaces of LaH⁺ and LaH²⁺ *J. Chem. Phys.* **94** 3722–9
- [77] Glenewinkel-Meyer Th, Kowalski A, Müller B, Ottinger C and Breckenridge W H 1988 Emission-spectra and electronic-structure of group IIIA monohalide cations *J. Chem. Phys.* **89** 7112–25
- [78] Glenewinkel-Meyer Th, Müller B, Ottinger C, Rosmus P, Knowles P J and Werner H J 1991 *Ab initio* calculations on the four lowest electronic states of AlF⁺ and AlCl⁺ *J. Chem. Phys.* **95** 5133–41
- [79] Mochizuki Y and Tanaka K 1999 Theoretical investigation of the GaF molecule and its positive ion *Theor. Chem. Acc.* **101** 257–61
- [80] Yoshikawa M and Hirst D M 1995 A theoretical study of the low-lying doublet states of the molecular ions GaF⁺ and GaCl⁺ *Chem. Phys. Lett.* **244** 258–62
- [81] Mochizuki Y and Tanaka K 1998 Theoretical investigation on the GaH molecule and its positive ion *Theor. Chem. Acc.* **99** 88–94
- [82] Boldyrev A I, Simons J and Schleyer P V 1993 *Ab initio* study of the electronic structures of lithium containing diatomic molecules and ions *J. Chem. Phys.* **99** 8793–804
- [83] Cao Z X, Wu W and Zhang Q E 1998 Spectroscopic constants and bonding features of the low-lying states of LiB and LiB⁺: comparative study of VBSCF and MO theory *Int. J. Quantum Chem.* **70** 283–90

- [84] Korek M, Rida M and Jbara A 2008 Theoretical calculation of the low lying electronic states of the molecular ion KH^+ *J. Mol. Struct.* **870** 100–5
- [85] Magnier S 2005 Theoretical description of the electronic structure of the alkali hydride cation NaH^+ *J. Phys. Chem. A* **109** 5411–4
- [86] Radzig A A and Smirnov B M 1985 *Reference Data on Atoms, Molecules, and Ions (Springer Series in Chemical Physics)* (Berlin: Springer)
- [87] Chanda A, Ho W C and Ozier I 1995 Hyperfine-resolved rovibrational spectrum of the $X^2\Pi$ state of HI^+ *J. Chem. Phys.* **102** 8725–35
- [88] Kardahakis S, Koukounas C and Mavridis A 2005 First principles study of the diatomic charged fluorides $\text{MF}^{+/-}$, $\text{M} = \text{Sc, Ti, V, Cr, and Mn}$ *J. Chem. Phys.* **122** 054312